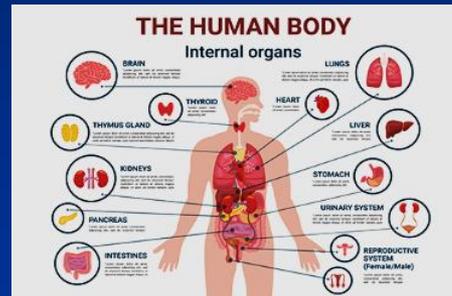




# 10. Acid-Base balance.

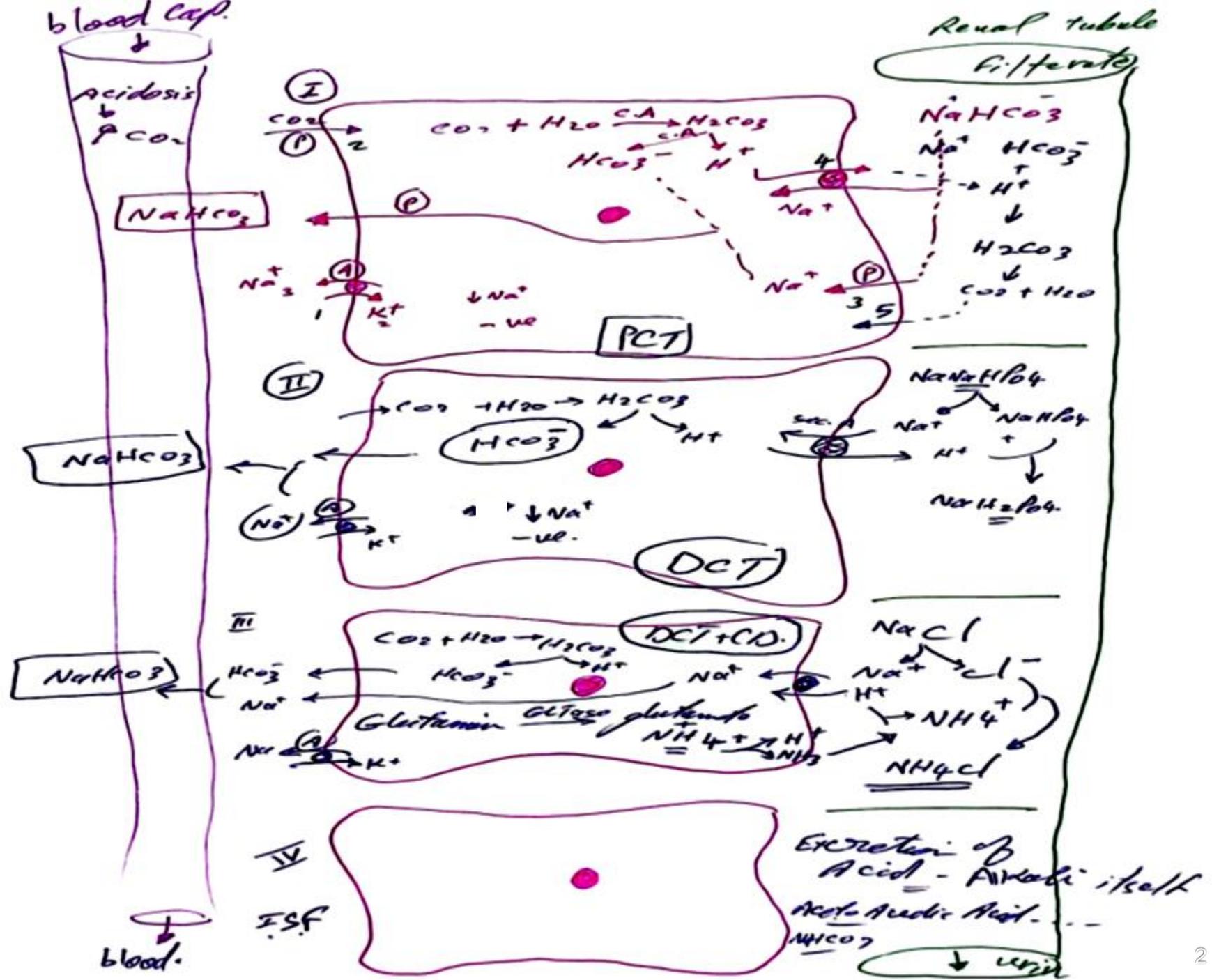


Sherif W. Mansour

Prof. of physiology

Mu'tah School of medicine

2024



# III. Renal Handling of Hydrogen & Bicarbonate

Bicarbonate  $\text{HCO}_3^-$  is the normal alkali reserve in the body which must be kept constant at a concentration of 26-28 mEq/litre at arterial  $\text{P}_{\text{CO}_2} = 40\text{mmHg}$ .

## Renal handling of $\text{HCO}_3^-$ :

### -In the PCT

More than 99% of  $\text{HCO}_3^-$  is reabsorbed by the kidney especially in PCT.

This reabsorption is affected by the acid base balance:

**Alkalosis** → increase excretion of  $\text{HCO}_3^-$  & urine becomes alkaline (normal urine is acidic  $\text{PH}=6$ )

**Acidosis** → complete  $\text{HCO}_3^-$  reabsorption.

Renal tubules are poorly-permeable to  $\text{HCO}_3^-$ . However, *it is reabsorbed in the form of  $\text{CO}_2$*  (highly permeable)

## Mechanism

**-Step 1 (intracellular):**  $\text{CO}_2$  of blood & of tubular fluid diffuses into cells of PCT  $\rightarrow$   $\text{CO}_2$  bind with  $\text{H}_2\text{O}$  in presence of Carbonic Anhydrase, CA enzyme  $\rightarrow$   $\text{H}_2\text{CO}_3 \rightarrow$  *ionized forming  $\text{HCO}_3^- + \text{H}^+$ .*

**-Step 2:**  $\text{H}^+$  is secreted to the lumen in *exchange with  $\text{Na}^+$*  (from filtered  $\text{NaHCO}_3$ ) by secondary active transport ( $\text{Na}^+/\text{H}^+$  counter-transport) & then  $\text{Na}^+$  diffuses to blood.

**-Step 3 (intra-luminal):** Secreted  $\text{H}^+$  combine with the **filtered  $\text{HCO}_3^-$**  by the help of CA enzyme in the brush luminal border of the PCT cells  $\rightarrow$   $\text{H}_2\text{CO}_3 \rightarrow$   $\text{H}_2\text{O}$  &  $\text{CO}_2$  (which diffuses into the tubular cells again ).

**-Step 4:** Formed  $\text{HCO}_3^-$  inside the cell moves passively through basal border to interstitium  $\rightarrow$  bind  $\text{Na}^+ \rightarrow$   $\text{NaHCO}_3$ .

## In late DCT & collecting ducts:

-In the intercalated cells of DCT & collecting duct,  $H^+$  secretion mechanisms is either dependent on  $Na^+$  (*formation of acid phosphate*) or independent of  $Na^+$  (*formation of ammonia*).

## Formation of acid phosphate:

**Step 1:** In tubular cells  $CO_2$  enters from blood inside the cell to  $CO_2 + H_2O + CA \rightarrow H_2CO_3 \rightarrow H^+ + HCO_3^-$ .

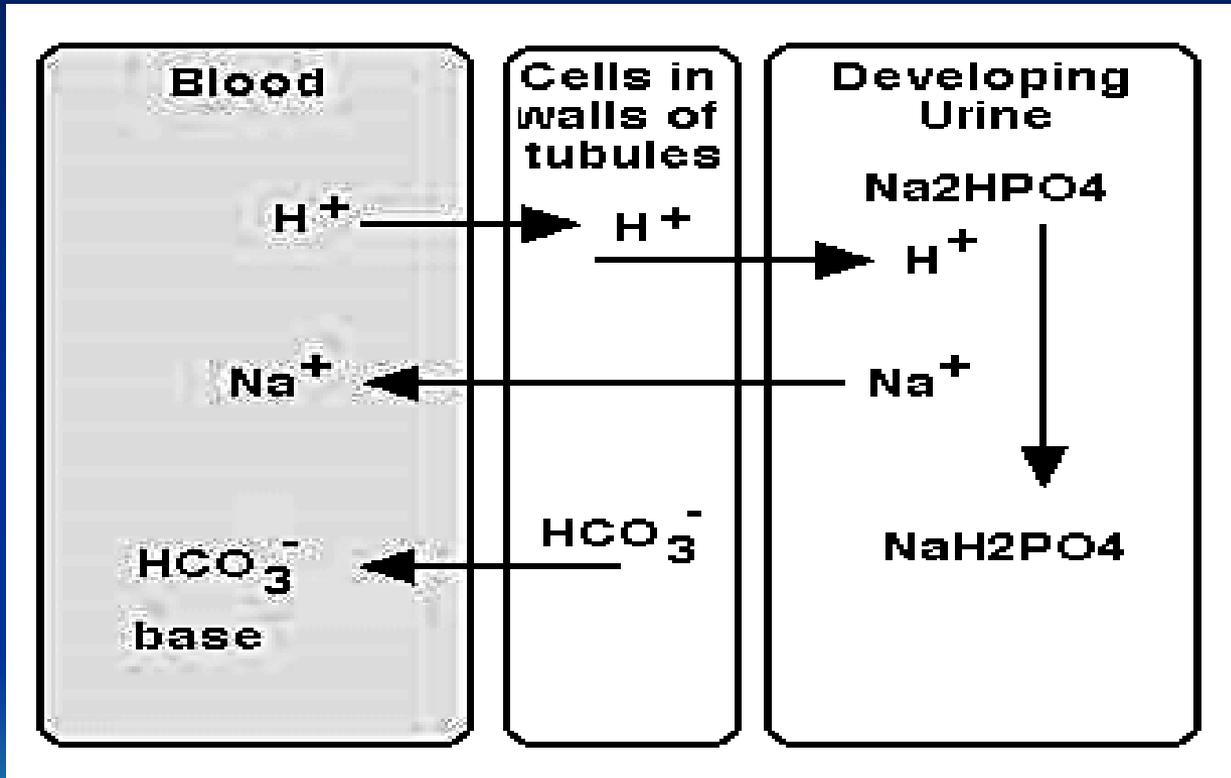
**Step 2:**  $H^+$  is secreted to the lumen in *exchange with  $Na^+$*  (from filtered  $Na_2HPO_4$ ) by *secondary active transport* ( $Na^+/H^+$  counter-transport) at the luminal border.

Then  $Na^+$  actively transported by to interstitial fluid by  $Na^+/K^+$  counter-transport through the basal membrane then to blood.

**Step 3:** In the lumen:  $H^+ + Na_2HPO_4$  (dibasic phosphate, alkaline)  $\rightarrow Na^+ + NaH_2PO_4$  (monobasic phosphate, acid).

**Step 4:**  $HCO_3^-$  formed inside the cell moves passively to interstitial fluid where bind with actively pumped  $Na^+$  to form  $NaHCO_3$  (alkaline tide) in blood.

# Steps of acid phosphate formation



## b- Ammonium salts formation in the proximal & distal tubules:

### Steps:

1- Metabolic reactions in the renal tubular **cells** produce  $\text{NH}_4^+$



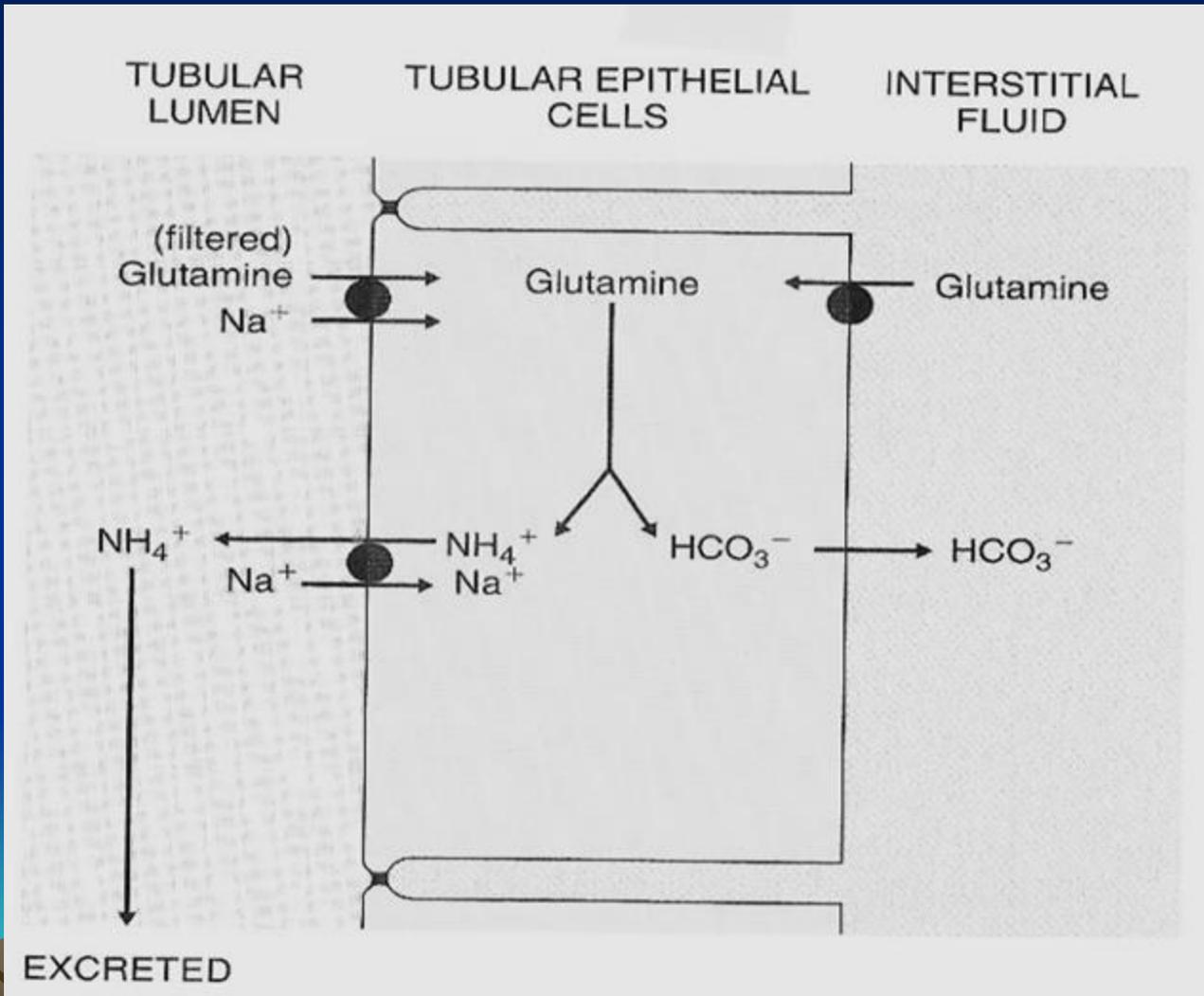
2-Alpha-ketoglutarate utilizes  $2\text{H}^+$  freeing  $2\text{HCO}_3^-$  And  $\text{NH}_4^+$  is in equilibrium with  $\text{NH}_3$  (ammonia) +  $\text{H}^+$

3- $\text{H}^+$  is secreted to the **lumen** by active proton (ATP-driven) pump present on the intercalated cells .However, ammonia (not ammonium) diffuse rapidly (non ionic diffusion), because it is lipid soluble, into tubular fluid (if urine is acidic)or diffuse to blood (if urine is alkaline).

**4-In the lumen:** Ammonia ( $\text{NH}_3$ ) +  $\text{H}^+$  (secreted) reconstructed to ammonium ( $\text{NH}_4^+$ ) which is trapped in the lumen and not re-diffused to blood (ammonia trapped mechanism). It is estimated that about 40mEq of  $\text{H}^+$  ion combines with  $\text{NH}_3$  are excreted/day.

⇒5-Then ammonium (lipid insoluble) combine with any anion ( $\text{A}^-$ ) either  $\text{Cl}^-$  of  $\text{NaCl}$  to form  $\text{Na}^+$  cation and  $\text{NH}_4\text{Cl}$  ammonium chloride salt or Phosphate group( of  $\text{Na H}_2\text{PO}_4$  to form  $\text{Na}^+$  cation and ammonium phosphate salt (  $\text{NH}_4\text{H}_2\text{PO}_4$ ). Both  $\text{NH}_4\text{Cl}$  and  $\text{NH}_4\text{H}_2\text{PO}_4$  are excreted in urine. However,  $\text{Na}^+$  is reabsorbed to blood to give  $\text{NaHCO}_3$  .so for each  $\text{H}^+$  ion excreted also one  $\text{NaHCO}_3$  is added to blood giving alkaline tide.

6-The amount of ammonium salts excreted in urine depends on activity of **glutaminase enzyme** &  $\text{H}^+$  secretion that increased in cases of chronic acidosis.



# Thank You

