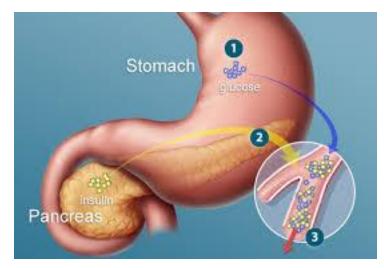
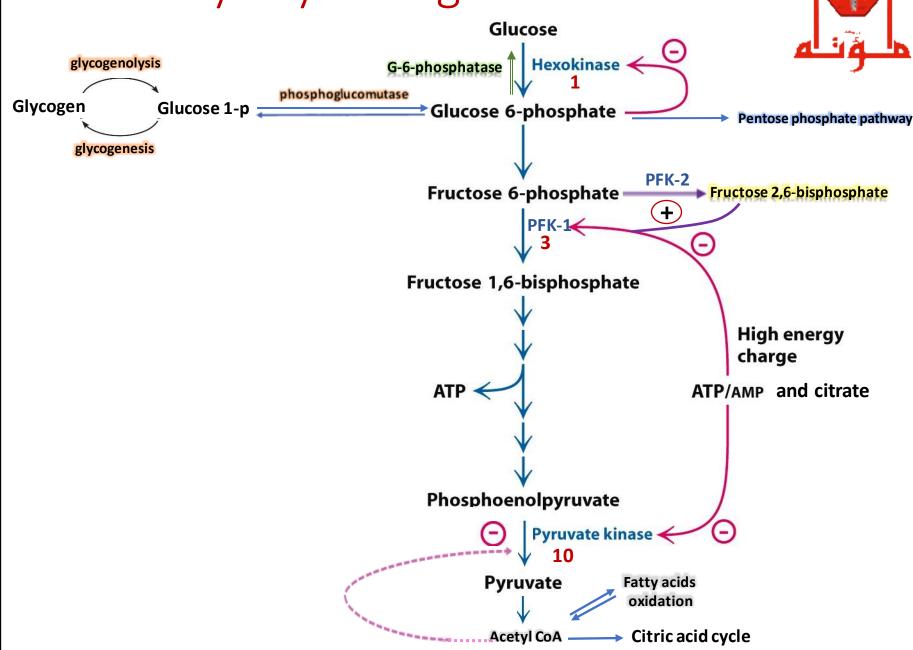


# Glycolysis II



Dr. Nesrin Mwafi
Biochemistry & Molecular Biology Department
Faculty of Medicine, Mutah University

## Glycolysis Regulation



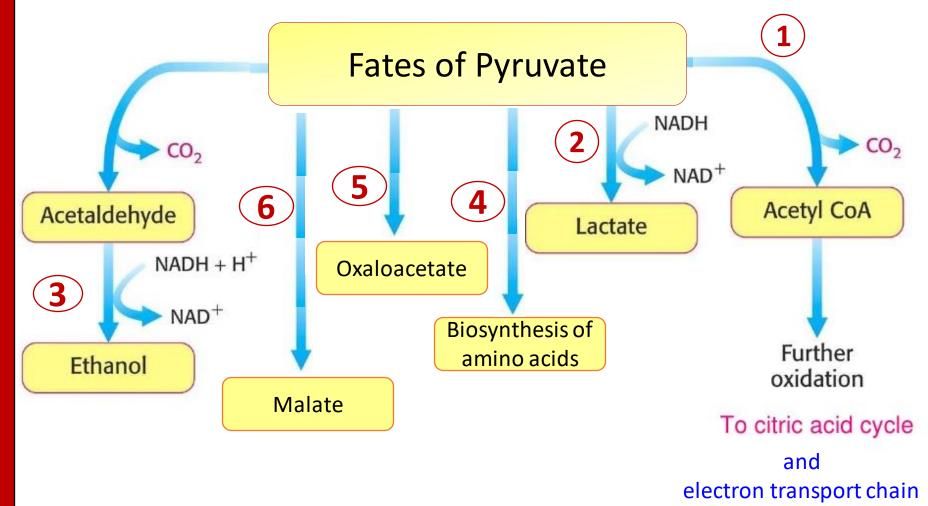
#### Glycolysis Regulation

القالم

- Glycolysis can be controlled at 3 points:
- Step 1 which is catalyzed by hexokinase enzyme (allosteric enzyme). Hexokinase isoforms (except glucokinase) are allosterically inhibited by excess G6P
- 2. Step 3 which is catalyzed by phosphofructokinase-1 enzyme. It is an allosteric enzyme. Two inhibitors are citrate and ATP whereas AMP and recently fructose 2,6-biphosphate (in liver) are activators. Actually this is the most important control point and it is considered as the main rate-limiting step in glycolysis
- 3. Step 10 which is catalyzed by pyruvate kinase enzyme. It is controlled by the level of ATP and Acetyl CoA (both are allosteric inhibitors). Accumulated Acetyl CoA in the cytosol is an indicator that the energy is now available from fat breakdown so no need to proceed in glycolysis

# Metabolic Fates of Pyruvate





# Metabolic Fates of Pyruvate

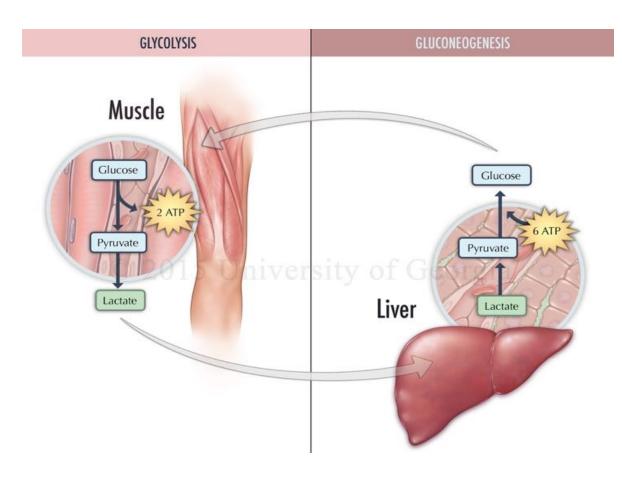


2. Lactic Acid Fermentation: bacteria, RBCs and O<sub>2</sub>-starved muscle cells

#### Cori Cycle

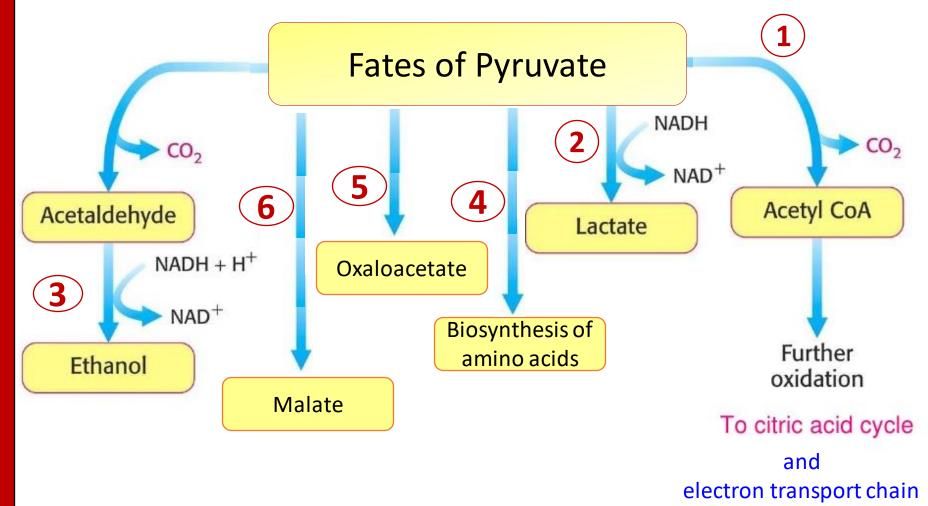
 Cori cycle "lactic acid cycle" is the metabolic pathway in which lactic acid produced in muscles during the time of oxygen depletion is converted back to glucose in the

liver by gluconeogenesis process



# Metabolic Fates of Pyruvate





- Metabolic Fates of Pyruvate
- 1. In aerobic conditions, pyruvate is converted to acetyl CoA which enters the citric acid cycle for further oxidation to CO<sub>2</sub> followed by oxidative phosphorylation
- 2. In anaerobic conditions like in lactic acid bacteria and some human cells (e.g. RBCs and O<sub>2</sub>-starved muscle cells), pyruvate is reduced to lactic acid with the concomitant oxidation of NADH to NAD+ (lactic acid fermentation)
- 3. In anaerobic conditions like in some M.O's (e.g. yeast), pyruvate is converted to ethanol
- 4. Amino acid biosynthesis: pyruvate is a precursor of some amino acids like alanine
- 5. & 6. Pyruvate can be used for synthesis of oxaloacetate or malate (both are TCA cycle intermediates)

#### Fluoride as Inhibitor of Enolase

- Oral bacteria depends on the food debris or dietary sugars found on the tooth surface as a primary source of energy. Acids are produced through fermentation process (harmful)
- Fluoride is a competitive inhibitor of enolase enzyme catalyzing Step 9
- Drinking fluoridated water or using a toothpaste containing fluoride inhibit the oral bacteria enolase activity. Consequently, this disrupts the bacteria glycolytic pathway and prevents formation of dental caries

#### Fluoride as Inhibitor of Enolase

- Sodium fluoride is known to have antiglycolytic effect that inhibits glycolysis by erythrocytes
- NaF tubes (gray top) are widely used for blood collection for glucose measurement
- Fluoride-containing tubes are suitable for blood collection if there is a long delay in blood separation following collection (false negative result)



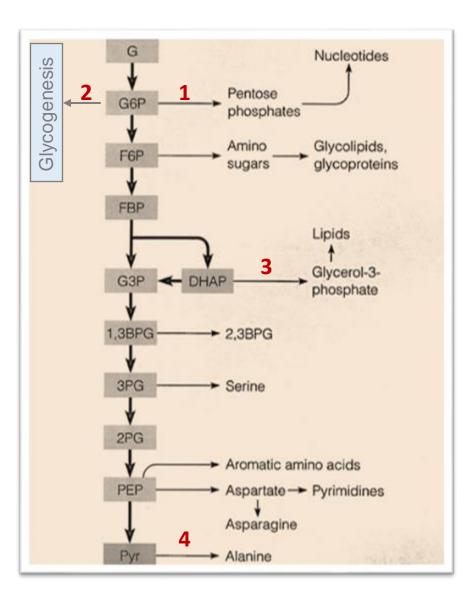
#### Glycolysis as Anabolic Pathway

- Glycolysis acts as catabolic as well as anabolic pathway. Therefore, glycolysis is very important central metabolic pathway
- Glycolysis intermediates with biosynthetic roles:
- Nucleotides biosynthesis: G6P is an initial substrate in pentose phosphate pathway (metabolic pathway which generates pentoses)
- 2. Glycogenesis via G6P
- 3. Lipids biosynthesis: DHAP is converted to glycerol-3-P
- Amino acids biosynthesis: pyruvate as precursor of alanine

### Glycolysis as Anabolic Pathway

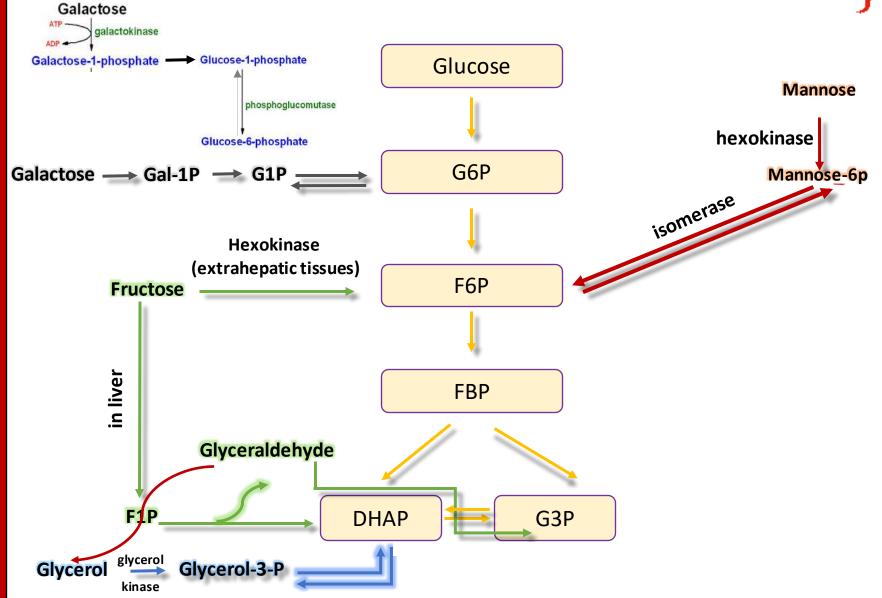


Major alternative fates of glycolytic intermediates in biosynthetic pathways



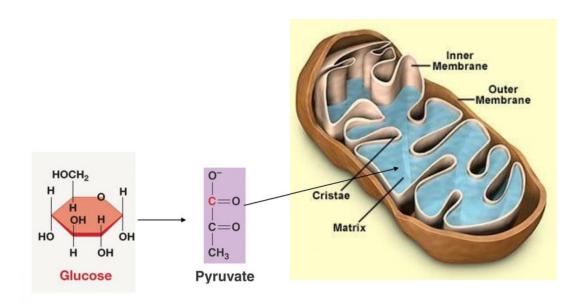
#### Other substrates enter Glycolysis





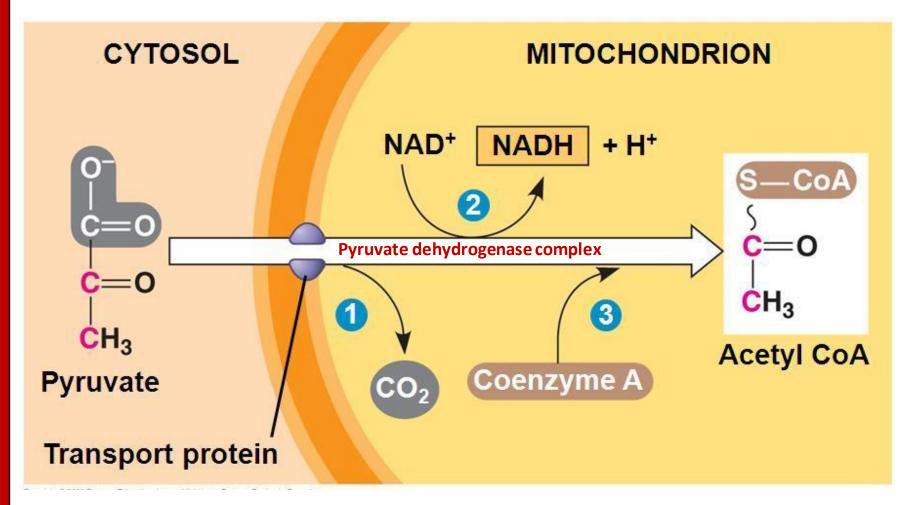
#### Acetyl CoA Formation

- In aerobic respiration, pyruvate (3C) joins the citric acid cycle after its conversion to acetyl CoA (2C)
- Citric acid cycle occurs in the mitochondrial matrix.
   Shuttling of pyruvate from the cytosol is facilitated by a transporter protein embedded in the inner mitochondrial membrane called pyruvate translocase



### Acetyl CoA Formation





#### **Acetyl CoA Formation**

- Pyruvate dehydrogenase complex (PDC) catalyzes
   the irreversible oxidative decarboxylation of pyruvate
   into Acetyl CoA with the release of CO<sub>2</sub>
- Energy-rich molecule "NADH" is also produced
- Coenzyme A (CoA) acts as acetyl group carrier due to its free sulfhydryl (–SH) end capable of forming thioester bond
- PDC is a multi-enzyme system consists of three catalytic enzymes and five coenzymes (three of them are prosthetic groups as they are tightly bound to their corresponding enzymes)

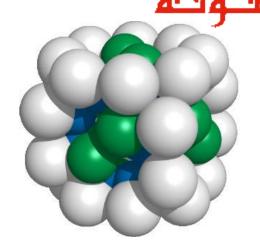
### Pyruvate Dehydrogenase Complex

- E1: pyruvate dehydrogenase
- E2: dihydrolipoamide transacetylase
- E3: dihydrolipoamide dehydrogenase

#### Coenzymes



- Lipoic acid (lipoamide) a prosthetic group of E2
- Flavin adenine dinucleotide (FAD) a prosthetic group of E3
- Coenzyme A (CoA or CoA-SH)
- Nicotinamide adenine dinucleotide (NAD+)

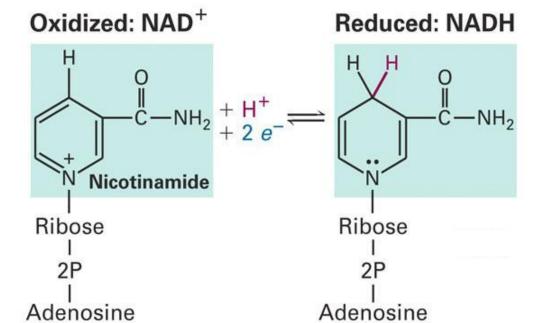


#### Coenzymes Structure

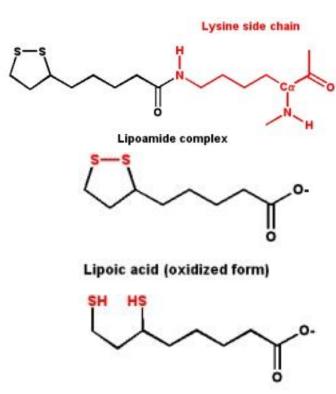


## Coenzymes Structure





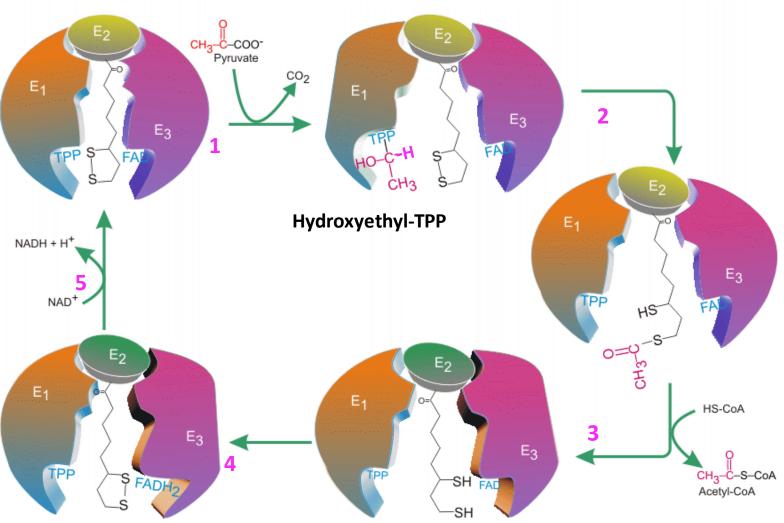
 $NAD^{+} + H^{+} + 2e^{-} \Longrightarrow NADH$ 



Lipoic acid (reduced form)

## Pyruvate Dehydrogenase Complex





#### Mechanism of PDC

- The mechanism by which this complex catalyzes the reaction is complicated but the main processes involve (5 steps):
- 1.Decarboxylation of pyruvate and the release of CO<sub>2</sub> a reaction catalyzed by E1-TPP. The product of this reaction "Hydroxyethyl moiety" is a substrate for the next reaction
- 2.The transfer of Hydroxyethyl moiety from TPP of E1 to lipoic acid of E2. This step is mediated by an oxidation of Hydroxyethyl to acetyl group coupled with reduction of disulfide bond
- 3. Transfer of acetyl group from lipoamide to CoA forming thioester bond and consequently Acetyl CoA is produced

#### Mechanism of PDC



- 4. Regeneration of disulfide bond of lipoamide via FAD (E3 prosthetic group) which is reduced to FADH2
- Regeneration of FAD by NAD+ which is reduced to NADH with the electrons transferred during the reaction (originally from Hydroxyethyl oxidation)