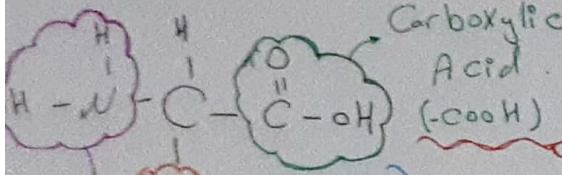


# Amino Acids

(300 Amino Acids)

1 The simplest Amino Acid is Glycine (R=H) → Key elements of amino acids: (C, N, O, H, S) bonded in chain group (R)



Amino Acids are Amphoteric molecules (Ampholytes) (have (-COOH) + (-NH<sub>2</sub>))

Chain group (R) → called (R) group

Amine Group (-NH<sub>2</sub>)

Amino Acids: are biologically important organic molecules that contain both Carboxylic Acid (-COOH) and Amine groups (-NH<sub>2</sub>).  
→ The basic structural building units of proteins + Energy Source.

Amino Acid Derivatives:

- 1 Tryptophan → Serotonin (Neurotransmitter)
- 2 Tyrosin → Thyroxin (Thyroid Hormone), Dopamine, Adrenaline, Noradrenaline (Catecholamine, Neurotransmitter)
- 3 Histamine → Histidine (Decarboxylation)
- 4 Glutamate → GABA (Inhibitor of WT, Neurotransmitter in brain)

\* Disulfide bond: Intramolecular (same Polypeptides) vs Intermolecular (Different Polypeptides)

Acidic Group (Proton Donor):  $\text{COOH} - \text{H}^+ \rightarrow \text{COO}^-$  (Anionic) Net charge (-1) (High pH)

Standard Amino Acids (Divided according to nutrition and Body requirement):

- Essential (not produced by the body) → Through Diet (8 Amino Acids)
- Non-Essential (Synthesized by body) → (9 Amino Acids)
- Conditionally essential (Synthesized by body)

Classification of Amino Acids:

- Five Amino Acids (5):
  - 3) positively (+) Basic: Lysine, Histidine, Arginine
  - 2) Negatively (-) Acidic: Aspartate, Glutamate
- Six Amino Acids (6):
  - 1) Carbon (2): α-Amino Acids, β-Amino Acids
  - 2) Carbon (3): Serine, Threonine, Cysteine (S group), Asparagine, Glutamine (-S-S bond)
  - 3) Carbon (4): Tyrosine (phenolic group)
  - 4) Carbon (5): Amino Acids

Essential (8 Amino Acids): Leucine, Lysine, Methionine, Tryptophan, Valine, Isoleucine, Phenylalanine, Threonine

Non-Essential (9 Amino Acids): Glycine, Alanine, Serine, Cysteine, Aspartic Acid, Glutamic Acid, Asparagine, Glutamine, Proline

Conditionally essential: It's Essential in Cases: Children, pregnant and Lactating women

Properties of Amino Acids:

- Non-polar: Glycine, Alanine, Valine, Leucine, Isoleucine, Methionine, Proline, Phenylalanine, Tryptophan (Hydrophobic)
- Charged Polar: Lysine, Histidine, Arginine
- Non-charged Polar: Serine, Threonine, Cysteine, Asparagine, Glutamine, Tyrosine (Hydrophilic)
- Acidic: Aspartate, Glutamate
- Basic: Lysine, Histidine, Arginine

Amino Group is Basic (Proton Acceptor):  $\text{NH}_2 + \text{H}^+ \rightarrow \text{NH}_3^+$  (Low pH)

Carbonic:  $\text{NH}_2 + \text{H}^+ \rightarrow \text{NH}_3^+$  Net charge (+1)

Non-Ionized:  $\text{NH}_3^+ - \text{CH}_2 - \text{COO}^-$  (pH = 7) Isoelectric point (pI) =  $\frac{1}{2}(\text{pK}_1 + \text{pK}_2)$

Both Essential and Non-Essential: Leucine, Lysine, Methionine, Tryptophan, Valine, Isoleucine, Phenylalanine, Threonine

It's Neutral (+0-) (net charge = zero)

Growth promoting factors, essential in growing child + pregnancy + Lactation

produced from Phenylalanine

# Amino Acids

→ Non proteinogenic + Non standard Amino Acids are derivative from (Standard Amino Acids).

Non-Standard (Non-Canonical), sometimes (Nonproteinogenic) (Post-translational modification) and

They are derivative from Standard Amino Acids

Standard (20)

\* We have (20) Standard (α-Amino Acids)  
\* And, (23) proteinogenic Amino Acids

Like (Ornithine) + (Citrulline) in the Urea cycle "part of Catabolism"

- 1) As intermediates during metabolism.
- 2) Posttranslational modification process.
- 3) Other enzymatic reactions.

(Canonical, natural, protein role → proteinogenic) always Adding in vivo protein synthesis (m-RNA translation)

All known as (α-Amino Acid) + primary Amino Acid exception (proline → secondary) Cyclic Pyrrolidine (side chain)

All of them Chiral + not superimposed

exception (Glycine → Achiral - superimposed) (Optical activity) X

Enantiomers: Two stereoisomers that are mirror images but not superimposable.

IS (Amino Group) on Right hand side → D (dexter) Left hand side → L (laevus) \* protein role (21)

Fischer projection: way commonly used to represent the structure of chiral molecules → Amino Acids → Carbohydrate. L-Amino Acids → Naturally Biologically isomer. (Bacterial Cell walls) → D-Amino Acids. → Methionine → Glycine (side chain) → Phenylalanine (Phenyl moiety) → Tryptophan

Optical Activity (Polarimeter) → used to Chirals clockwise (+) Counterclockwise (-)

Racemic mixture = equal amounts of each enantiomers (net = zero). → Tyrosine (phenolic group) → Cysteine (-S-S-) disulfide bond covalent linkage

20 standard 3 Non-standard

\* Natural (Amino Acids) (21) → (20) standard α-Amino Acid (β-peptides) (1) non standard β-Amino Acid (β-Alanine). \* Chiral Amino Acid (19) standard (- Glycine)

\* Primary α-Amino Acid (standard) (19) → standard (- proline).

\* protein role (21) (1) non standard (Hydroxyproline)

\* β-Amino Acids are non standard

\* β-peptides used to synthesis (Antibiotics)

\* Isomers = molecules with same molecular formula but different chemical structures. 1) Constitutional (structural) isomers (different functional group) 2) Stereoisomers (spatial isomers) Atomic connectivity

\* Post-Translational Modification (essential function → protein):

- 1) Carboxylation → Glutamate (Blood Clotting Cascade)
- 2) Hydroxylation → proline (in Collagen protein in connective tissue).
- 3) Phosphorylation (+ OH group) (Serine, Threonine + Tyrosine) negative charge → protein and Enzyme.
- 4) Glycosylation (add sugar) stabilizes protein + Direct selected protein → Intracellular organelles

Non proteinogenic + Non standard Amino Acids [may] → protein role → nonprotein role

They are active by themselves and have a function.

\* [Non proteinogenic] + Non standard Proteinogenic

↑ derivatives by

Standard Amino Acid

Carnitine and GABA → Non protein role in proteins non-proteinogenic

(Hydroxyproline) → not produced directly by standard cellular machinery. \* Hydroxyproline → Non standard non proteinogenic protein role