# Archive <br> Subject 

## Collected by



## ARCHIVE MID BIOCHEM, WAREED

1-If the Normality=3 of the H 2 CO 4 ,the Mortality is?
a) 1.5
b) 1.8
c) 3
d) 2

## 2-If the Mortality $=\mathbf{6}$, the Normality of H2PO4 is?

a) 15
b) 19
c) 18
d) 14

## 3-Which of these is a polar covalent bond ?

a) $\mathrm{P}---\mathrm{Ca}$
b) $\mathrm{Na}----\mathrm{Cl}$
c) $\mathrm{H}------\mathrm{H}$
d) $\mathrm{H}----\mathrm{CL}$

4-One of the following is not a canonical amino acid property ?
a) L-amino acids only
b) D-amino acids only
c) All Alpha amino acids
d) None of these

## 5-A boy's body lacks collagen, which amino acid would makeup for the lack of the following ?

a) Glycine
b) prolin
c) Hydroxyproline
d) Isoleucine

## 6-(-NH)group is called?

a) amino group
b) hydroxyl group
c) carboxyl group

## 7-Ligases enzyme is?

a) join two molecules with covalent bond
b) cleaves various bonds
c) catalyzes isomerisation
d) transfer functional group

8-Which of the following is true about micropeppite ?
***they are subject to inaccuracies caused by the changing environment

## 9-How the polypeptide chain is formed ?

a) carboxyl group give - OH from amino acid, and amino group give -H from another amino acid
b) carboxyl group give -H from amino acid, and amino group give -OH from another amino acid

10-If you know that the chemical formula of the amino acid glycine is $\mathbf{C 2 H 5 N O 2}$, then what is the molecular formula of a peptide that is composed out of 10 residues of glycine?
a) C 20 H 32 N 100 O 11
b) C 20 H 50 N 100 O 11
c) C 20 H 30 N 200 O 11
d) C 2 OH 32 N 100 O 15

## 11-Tripeptide chain?

a) 3 amino acid, 2 peptide bond
b) 2 amino acid , 1 peptide bond
c) 4 amino acid, 3 peptide bond
d) 5 amino acid, 4 peptide bond

12- The process of making the peptide bond ?
a) Translation
b) Dehydration synthesis
c) Transcription

13-Which one is non-essential amino acid?
a) leucine
b) valine
c) tryptophan
d) glycine

## 14-The peptide bond ?

a) Lost $(-\mathrm{OH})$ from carboxylic group and lost (-H) from amino acid
b) Lost $(-\mathrm{H})$ from carboxylic group and lost $(-\mathrm{OH})$ from amino acid

15-The reason for the formation of phenylketonuria?
***Defect in the enzyme
16-what is the biochemical laboratory test in relation to assisting in the early diagnosis of disease ?
a) Creatine kinase (ck-MB)
b) ALT
c) TSH
d) CEA

17-Which of the following is correctly matched?
a) tristearin is a conjugated lipid
b) cholic acid is simple lipid
c) plasmalogen is derived lipid
d) estrogen is simple lipid

18-The order of the density of lipoprotein from lowest to highest ?
a) $\mathrm{CM}, \mathrm{VLDL}, \mathrm{LDL}, \mathrm{HDL}$
b) $\mathrm{HDL}, \mathrm{LDL}, \mathrm{VLDL}, \mathrm{CM}$
c) VLDL ,LDL ,HDL ,CM
d) $\mathrm{LDL}, \mathrm{CM}, \mathrm{HDL}, \mathrm{VLDL}$

19-Which of the following sciences are used to study the structures of genes?
a) glycomics and molecular genetics
b) molecular diagnostics and molecular biology

20-What is the essential amino acid for pregnancy and children ?
a) Arginin
b) Lysine
c) Isoleucine
d) glutamine

21-What of the following makes Vitamine D?
a) Decarboxylation-7
b) Hydroxylation-7

22-If the CO2 level increases in the blood, PH becomes
a) 7,4
b) 7,3
c) 7,6
d) 8,3

23- Which of the following are the correct abbreviations of these amino acids ,(Therionin ,Arginine ,Serine ,Glycine,Serine)?
a) $T, R, S, G, S$
b) $T, A, S, G, S$
c) $H, A, S, G, S$

24-Solution has $\mathrm{PH}=6,[\mathrm{~A}-]=4$, what is the PKa ?
a) 6
b) 1.8
c) 9
d) 3.2

## 25-What causes Alzheimer disease?

***accumulation of beta amyloid.
26- Which one of these is not an acidic amino acid, not positively charged ?
a) glutamate
b) glutamic acid
c) aspartate
d) aspartic acid
e) glutamine

27- One of these fatty acids has omega 3 in its structure ?
***Linolenic acid

## 28-What is Ceramide ?

***Sphingosine connected by an amide linkage to a fatty acid.

29- Which bond is found only in pleated sheets?
a) hydrogen bond
b) ionic bond

30-Given that for a glycine PK1=2.4 and PK2=9.8 , which of the following PH points is incorrectly matched with the charge of glycine at that point ?
a) $\mathrm{PH}=1$ +1
b) $\mathrm{PH}=13 \ldots \ldots \ldots \ldots \ldots-2$
c) $\mathrm{PH}=6.1 . \ldots \ldots \ldots \ldots . .$. . Zwitterion
d) $\mathrm{PH}=9.8 \ldots \ldots \ldots \ldots . . .50 \%$ neutral and $50 \%$ negative
e) $\mathrm{PH}=14 \ldots \ldots \ldots \ldots . . . .-1$

## GOOD LUCK

BATOOL FREHAT , Wateen

