

# Amino Acids 1



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### Amino Acid Structure



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- to R and called B and so on

- · Amino acids are biologically important organic molecules that contain both carboxylic acid (-COOH) as well as amine  $(-NH_2)$  groups
- The side-chain also called "R" group is specific to each amino acid



iological significance of Amino Acio 1. Amino acids are N-containing molecules 2. The basic structural building units (monomers) of proteins 3. Precursors of many biomolecules like neurotransmitters (non-protein role) 4. They are also utilized as an energy source • There are 20 standard (canonical) amino acids which are encoded directly by triplet codons in the universal genetic code during in vivo protein synthesis process (mRNA translation) stendard -> amino acid is incorporated to the pulypetical chain son MANA translation process



Ala

Val

Glu

3rd base in codor

G

Glv



# Standard Amino Acids List

	, tio	
Histidine	Arginine	Alanine
Isoleucine	Asparagine	Asparatate
Leucine	Glutamine	Cysteine
Methionine	Glycine	Glutamate
Phenylalanine	Proline	
Threonine	Serine	
Tryptophan	Tyrosine	
Valine		
Lysine		

Standard Amino Acids They are proteinogenic and natural amino acids 1. (the other proteinogenic amino acids N-formyl methionine, pyrrolysine and selenocysteine are called non-standard or non-canonical amino acids) protein ogenic fictus stem loop structure and wonstander of it's with "SECIS element" فريد محافوند but they are add by unique Eukaryotes Synthetic mechanismi Lingth automatics (ب حر ملوب حو هم الطريقة) به نعرك مبضاى بطرة ما يرة حي المعلى 5mmmuganna 3mm mRNA 3' UTR

Incorporation of selenocysteine in protein structure by unique mechanism



### Standard Amino Acids



3. They are all chiral molecules (except glycine which has achiral C) with L- stereochemical configuration (left-handed isomers)



### Standard Amino Acids



- Chiral molecules should contain at least one chiral center (usually a carbon atom)
- Chiral carbon: asymmetric carbon atom attached to 4 different groups of atoms





### Isomerization



- Isomers: are molecules with same molecular formula but different chemical structures
  - 1. Constitutional (structural) isomers: atoms and functional groups bind together in different ways
  - 2. Stereoisomers (spatial isomers): differ in the configuration of atoms rather than the order of atomic connectivity

## D/L Amino Acids



- Enantiomers: are two stereoisomers that are mirror images to each other but not superimposable
- D- (dexter)/L- (laevus) Nomenclature system: commonly used to assign the configurations in sugars (carbohydrates) and amino acids
- As a rule of thumb: if the amino group is on the right-hand side of α-carbon at Fisher projection, the configuration is D. If it is on the left-hand side, the configuration is assigned as L.



## D/L Amino Acids



- Most naturally occurring sugars are D-isomers while most naturally occurring amino acids are Lisomers (amino acids of protein)
- D-amino acids polypeptides (right-handed isomers) are components of bacterial cell walls to resist digestion by other organisms



Optical Activity



- (+)/(-) nomenclature system: if one enantiomer rotates the light clockwise, it is labeled (+) or (a) (dextrorotatory), The second mirror image enantiomer is labeled (-) or (l) laevorotatory
- D/L system should not be confused with +/- or d/l system. For example, D-isomer might be levorotatory
   9 of 19 L-amino acids commonly found in proteins are dextrorotatory 10-11, 3, 12, 10
- Racemic mixture contains equal amounts of each enantiomer (net rotation is zero)



## Classification of Amino Acids

- $\beta$ -amino acids are non-proteinogenic with  $\beta$ -alanine is the only common naturally occurring  $\beta$ -amino acid,  $\beta$ -alanine is used in plants and microorganisms in the synthesis of pantothenic acid (vitamin  $B_5$ )
- Unlike  $\alpha$ -peptides, The  $\beta$ -peptides are artificial peptides used in some antibiotics to counter resistance as they are more stable against proteolytic degradation



# ategories of Standard Amino Acids

- The 20 standard amino acids are classified into 3 major categories according to the polarities of their "R" groups:
  - 1) Amino acids with non-polar R groups
  - 2) Amino acids with charged polar R groups
  - 3) Amino acids with uncharged polar R groups



#### Amino acids with non-polar R groups



- · Glycine has the simplest side chain: H atom
- Alanine, valine, leucine and isoleucine have aliphatic hydrocarbon side chains
- Methionine has a thioether side chain (sulfur atom)
- · Proline has a cyclic <u>pyrrolidine</u> side chain
- · Phenylalanine has a phenyl moiety
- Tryptophan has an indole group

#### Amino acids with charged polar R groups



3 amino acids are positively charged (basic) and
 2 amino acids are negatively charged (acidic)



#### Amino acids with charged polar R groups

**آ** 

- · Arginine has a guanidine group
- · Lysine has a butyl ammonium side chain
- · Histidine has imidazole group
- Aspartic and glutamic acids in their ionized state are called aspartate and glutamate, respectively



6 amino acids with hydroxyl, amide or <u>thiol</u> groups





- + ryrosive • Serine and threenine bear hydroxyl (-OH) R group
- · Asparagine and glutamine have amide bearing side chains. They are the amide derivatives of aspartic and glutamic acids - wind group <sup>جاري</sup> باجليتة
- Tyrosine is aromatic and has a phenolic group
- · Cysteine has a thiol group that can form a disulfide bond (-S-S-) with another cysteine through the oxidation of 2 thiol groups (cystine is the oxidized dimeric form). The disulfide bridge in proteins contributes to the stability and overall shape of a protein



#### Amino acids with uncharged polar R groups



- Cysteine residues may be separated from each other by many amino acids in the primary sequence of a polypeptide or may even be located on two different polypeptides. The folding of the polypeptide chain(s) brings the cysteine residues into proximity and permits covalent bonding of their side chains.
- Disulfide bond could be intranolecular (2 cysteine residues on the same polypeptide chain) or intermolecular (2 cysteine residues on two separate/ different polypeptide chains)

### Amino Acids Abbreviations

	/mare comone	- They t	he Thether
3 121	3-letters	1-letter	Amino acid
	Ala	А	Alanine
	Arg	R	Arginine the met
	Asn	N	Asparagine
	Asp	D	Aspartic acid (Aspartate)
	Cys	С	Cysteine
	Gln	Q	Glutamine
	Glu	Е	Glutamic acid (Glutamate)
	Gly	G	Glycine
	His	н	Histidine
	Ile	I	Isoleucine
	Leu	L	Leucine
	Lys	К	Lysine
	Met	М	Methionine
	Phe	F	Phenylalanine
	Pro	Р	Proline
	Ser	S	Serine
	Thr	т	Threonine
	Trp	W	Tryptophan
	Tyr	Y	Tyrosine
	Val	v	Valine



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