

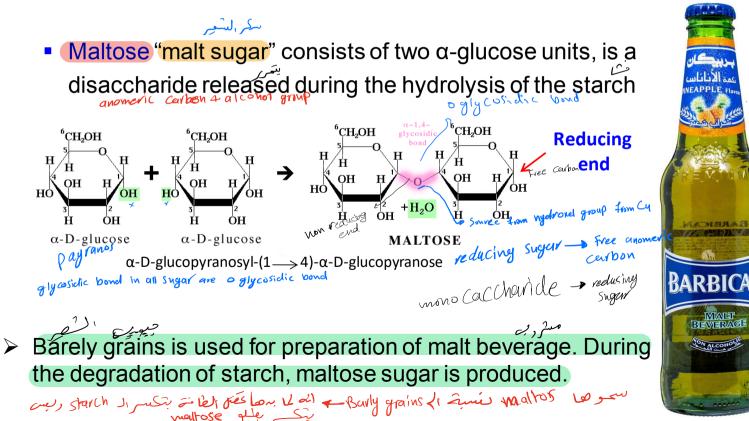
Carbohydrates



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These are two monosaccharides linked together via the glycosidic bond. Three common disaccharides:



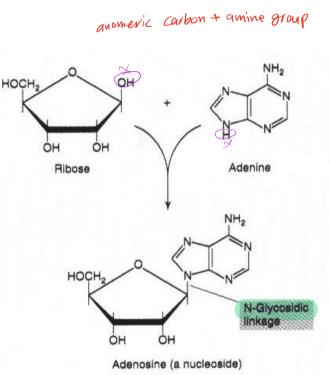
Glycosidic bond



□ Glycosidic bond is a type of covalent bond where the anomeric group of a sugar can condense with an alcohol. This type of bond is called O-glycosidic bond.

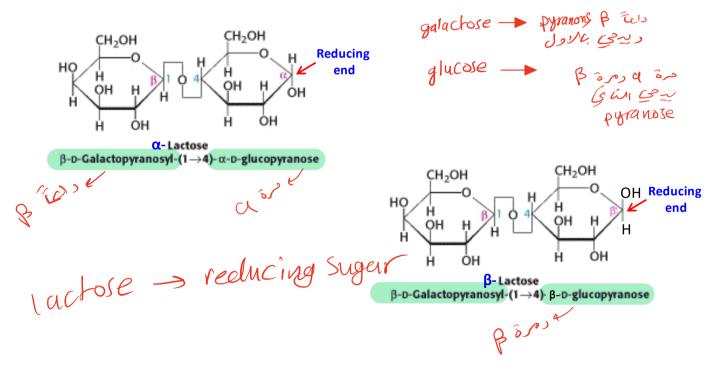
N-glycosidic bond is another type of glycosidic bond which forms between the anomeric carbon of sugar and an amine.

e.g. the bonds that link D-ribose and D-deoxyribose to purines and pyrimidines in the nucleic acids: RNA & DNA, respectively.





 Lactose "milk sugar" consists of glucose & galactose, is a disaccharide occurs naturally in the milk (dairy products)



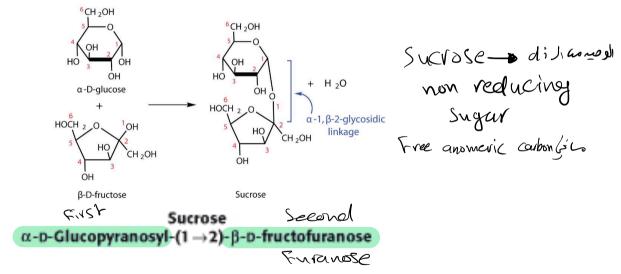


 Lactose Intolerance: deficiency of lactase enzyme leading to Gastrointestinal tract (GIT) disturbances such as: nausea, bloating, abdominal cramps and diarrhea due to digestion of lactose (intact) by bacteria found in colon

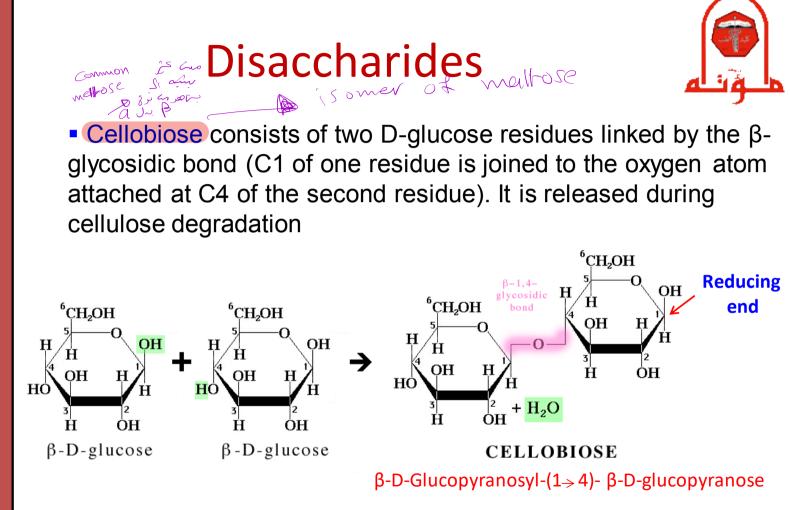




 Sucrose "table sugar" consists of glucose & fructose, is a disaccharide obtained commercially from cane or beet.



Sucrose is not a reducing sugar because the anomeric carbon of the second residue (the reducing end) is not free but involved in the glycosidic bond formation.



Cellobiose is an isomer of maltose (stereochemistry of the glycosidic bond which is β in cellobiose and α in maltose)

Polysaccharides Polymers

Polysaccharides "glycans" are polymeric molecules consist of long chains of monosaccharide units bound together via the glycosidic linkages.

□ Polysaccharides composed of same type of monosaccharides are called homopolysaccharides "homoglycans" and those consisting of more than one type are called heteropolysaccharides "heteroglycans".

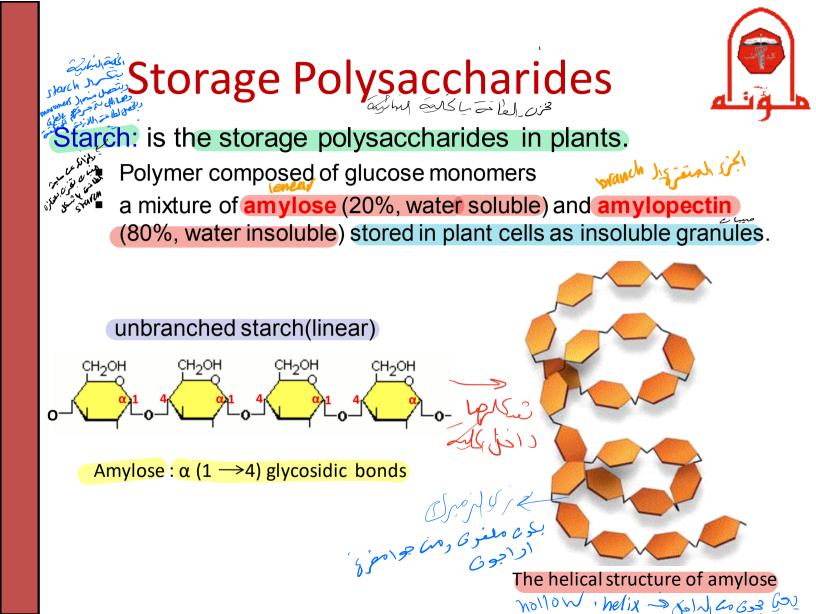
□ They form branched as well as linear polymers.

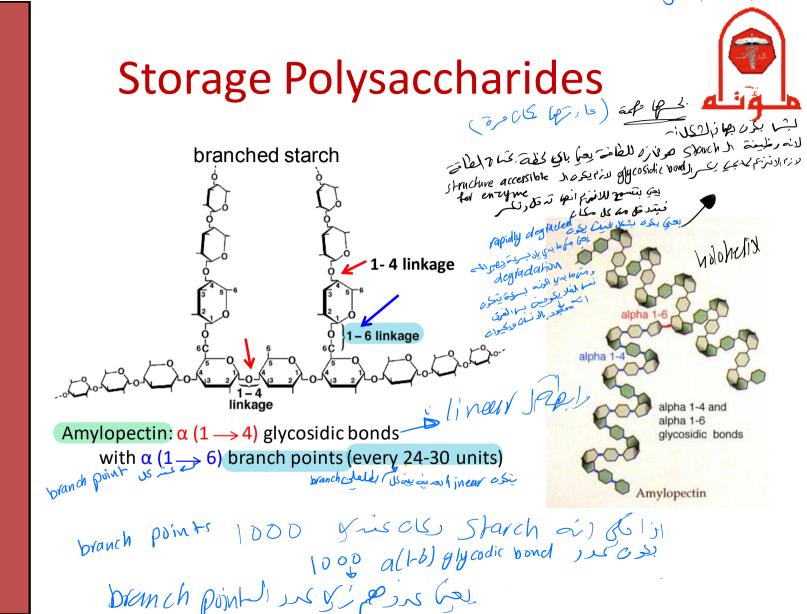
They are classified into: Function June

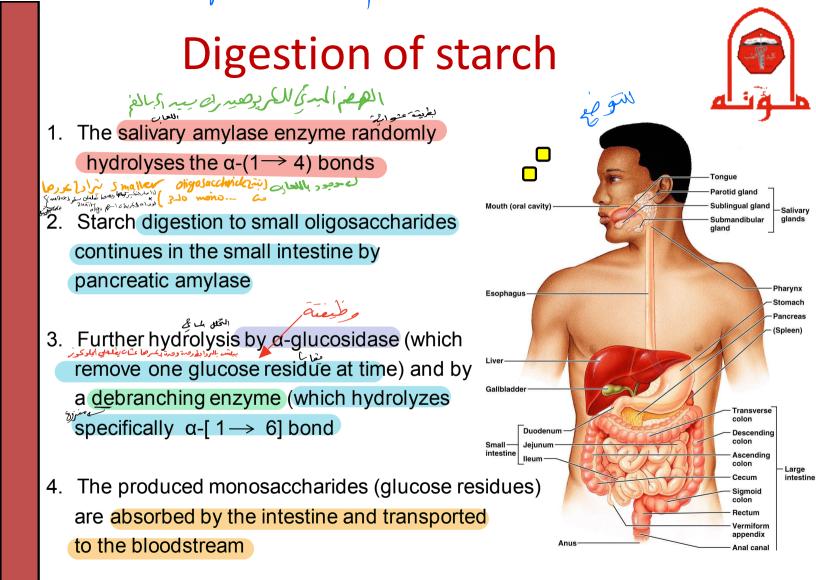
Storage polysaccharides like starch and glycogen

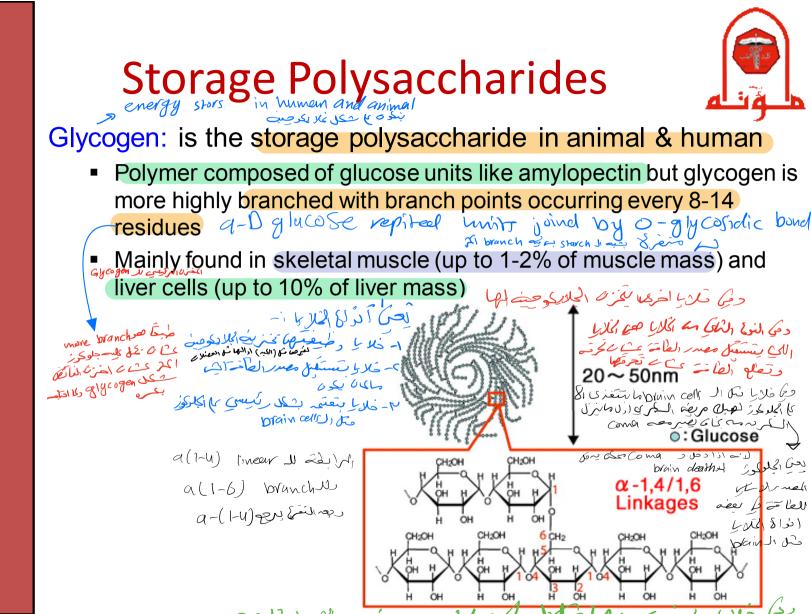
2. Structural polysaccharides like cellulose and chitin (1) and glad D

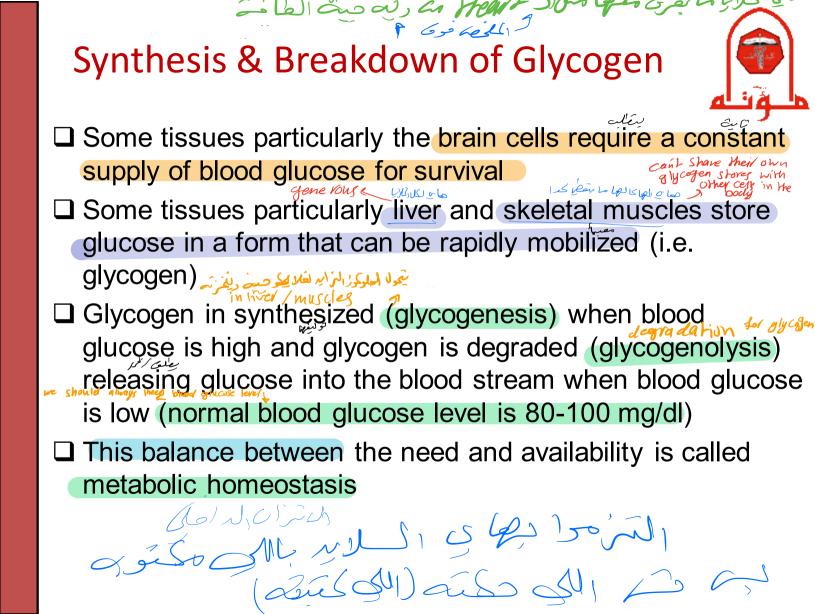
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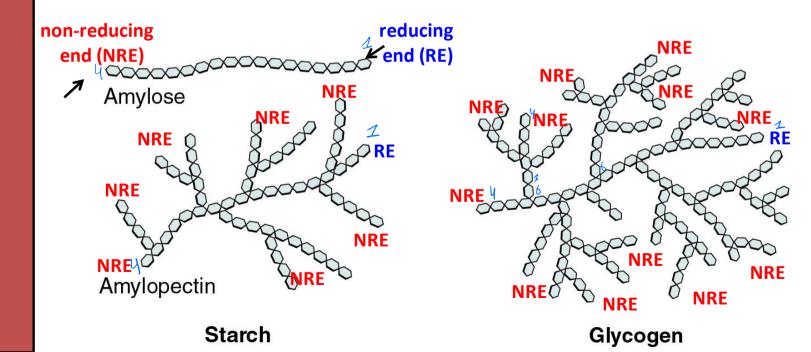




Storage Polysaccharides



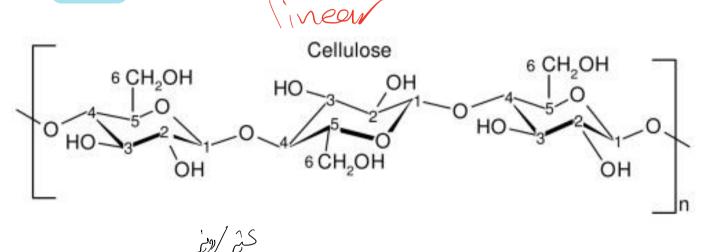
Starch and glycogen have one reducing end (the molecule end containing a free anomeric carbon C1). On the other hand, the branches ends are all called non-reducing ends and being sites where enzymatic lengthening and degradation occur.



Cellulose: the primary structural component of plant cell walls.

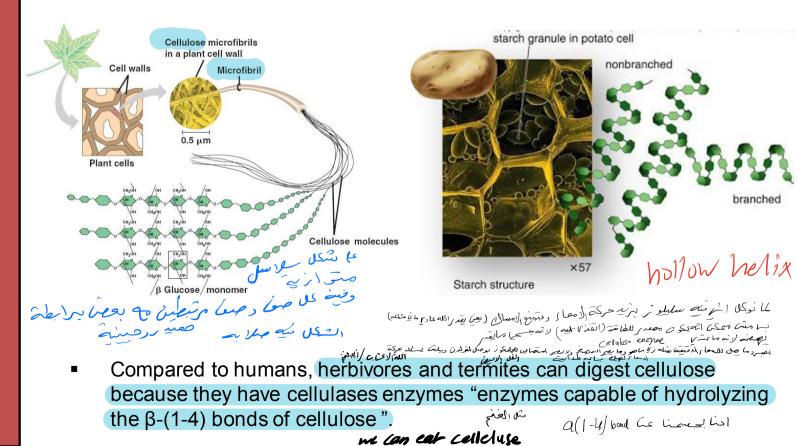
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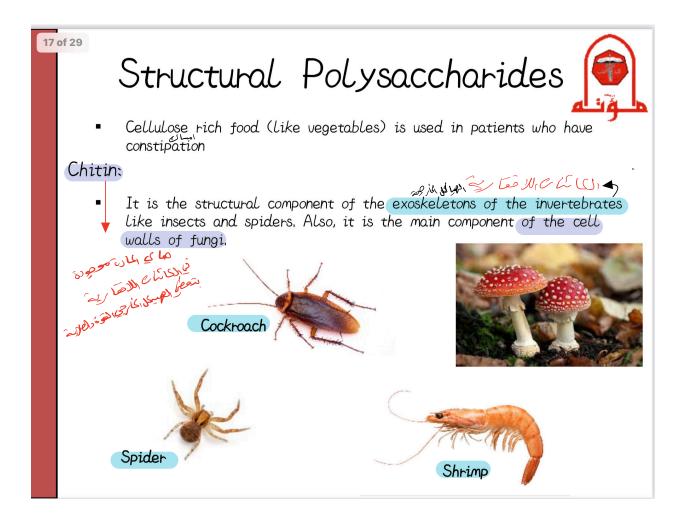
 A linear polymer of D-glucose residues linked via β-(1-4) glycosidic bonds.



- It is the most abundant organic molecule on the earth. cellulose accounts for over half of the carbon in the biosphere.
- It adopts a very different molecular architecture from that of starch (hollow helix) due to its β-linkages.

- Cellulose forms very long straight chains. The parallel chains interact with one another through H-bonds



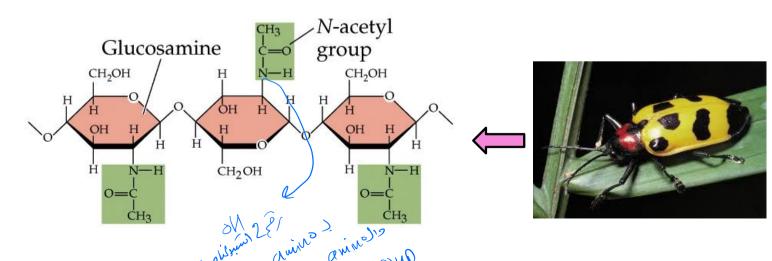




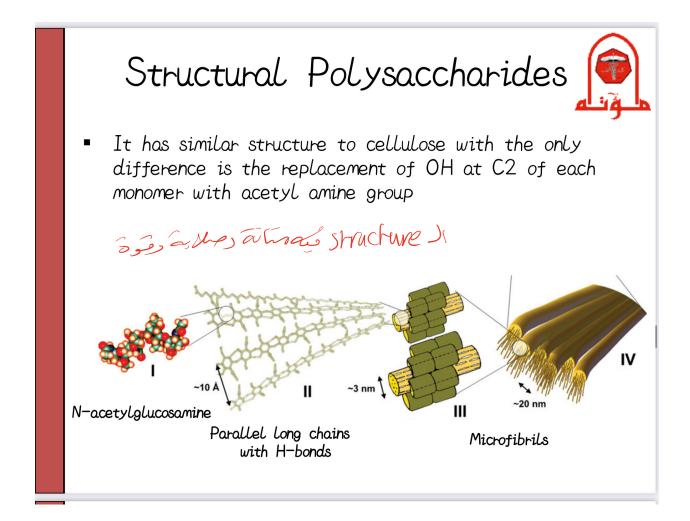
 Cellulose rich food (like vegetables) is used in patients who have constipation

Chitin:

- It is the structural component of the exoskeletons of the invertebrates like insects and spiders. Also, it is the main component of the cell walls of fungi.
- A long chain polymer of N-acetyl-D-glucosamine residues joined by β-(1-4) bonds.



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It has similar structure to cellulose with the only difference is the replacement of OH at C2 of each monomer with acetyl amine group
Chitosan: is a linear polysaccharide composed of randomly distributed by β-(1-4)-linked D-glucosamine (deacetylated unit) and N-acetyl-D-glucosamine (acetylated unit). It is produced commercially by deacetylation of chitin (e.g. by treating shrimp shells with the alkali sodium hydroxide).



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Medical uses: it is useful in weight loss and obesity treatment plans because it can reduce fat absorption + it has high fiber contact full besorb the water we then full closer by the water we then full





- Consist of two or more different monosaccharide units and are closely associated with lipid (glycolipids) or protein (glycoproteins)
- The naturally occurring heteroglycans are mostly found in the connective tissues (such as cartilage, tendon, blood vessel walls,.....etc)

1. Hyaluronic acid (Hyaluronate)

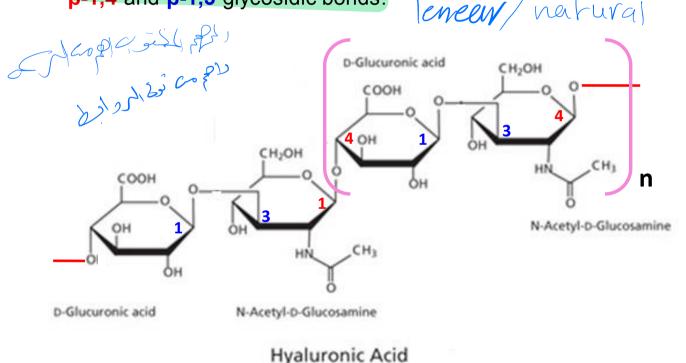
 It is the major component of joint fluid (syňovial fluid). It acts as a lubricating agent and shock absorber.

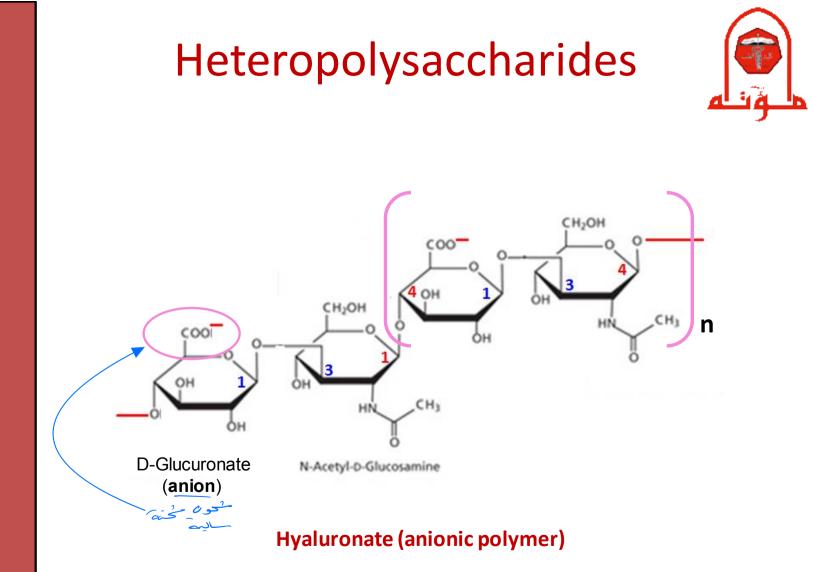


- It is also a major component of skin, where it is involved in tissue repair. Dry and scaly skin such as that caused by eczema may be treated with a prescription skin lotion
- containing sodium hyaluronate as its active ingredient. بر مبان الدرنين بر مبان الدرنين



- **آ**
- Hyaluronic acid is a **linear** polymer of the disaccharides " Dglucuronic acid and N-acetyl-D-glucosamine " linked via alternating β -1,4 and β -1,3 glycosidic bonds.

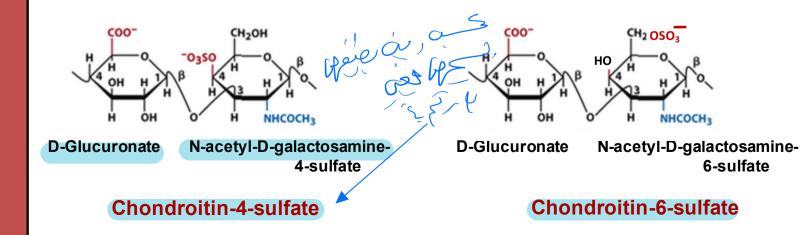




Heteropolysaccharides Sulfate group 2006 2000 Structure U = Court



- 2. Sulfated heteroglycans these consist of sulfated disaccharide units such as: chondroitin sulfate, dermatan sulfate, keratan sulfate and heparin
 - Chondroitin-4-sulfate & Chondroitin-6-sulfate are unbranched polymers containing the disaccharide "D-glucuronic acid and N-acetyl-Dgalactosamine " with the N-acetyl-D-galactosamine OH groups at position 4 and 6 being sulfated, respectively.



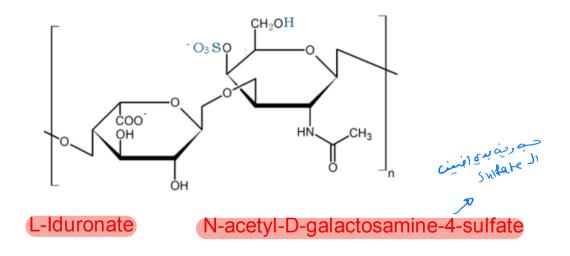


- Chondroitin sulfate is a major component of cartilages. They provide them with resistance to compression. Loss of chondroitin sulfate from the cartilage is a major cause of osteoarthritis.
- Chondroitin is used as dietary supplement to treat osteoarthritis. It is commonly sold together with glucosamine





 Dermatan sulfate: is a natural polysaccharide found mostly in the skin. It is a linear polymer of a disaccharide containing L-Iduronic acid (modified L-Idose sugar) and N-acetyI-D-galactosamine-4sulfate

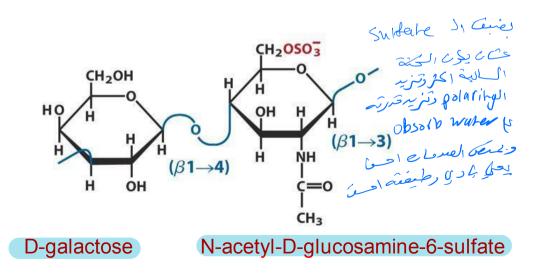


Dermatan sulfate

connective tyshell = 1 - 2 - 1 - 1 - 1 - 1 - 1 - 1



 Keratan sulfate: is a natural polysaccharide mainly found in the cartilage and bone. It is highly hydrated molecules which in joints can act as a cushion to absorb mechanical shock. This linear polymer is consisting of repeating disaccharide unit containing D-galactose and N-acetyl-D-glucosamine-6-sulfate



Keratan sulfate

Heteropolysaccharides the most nigh charg polemer polarity 2410



 Heparin: is the most highly charged polymer of any known biological molecule. Heparin is a complex mixture of linear polysaccharide and it varies in the degree of sulphation of its sugar units. One example is the sulfated disaccharide unit containing L-Iduronate-2-sulfate and N-sulfo-D-glucosamine-6-sulfate

